

# Raman and Infrared Spectroscopic Tentative Identification of Organic Traces in Sadiya (LL5) Ordinary Chondrite

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**Abstract.** We report here for the first time the possible presence of organic compounds in Sadiya (LL5) ordinary chondrite using micro-Raman and infrared spectroscopic technique. The micro-Raman spectrum exhibits the diamond and graphite peaks correspondingly at 1331  $\text{cm}^{-1}$ , 1349  $\text{cm}^{-1}$  and 1588 – 1618  $\text{cm}^{-1}$ . The full wave at half maximum value of about 18  $\text{cm}^{-1}$  for Sadiya indicating the nature of disordered phase involved shock metamorphism in the meteorite samples. The diamond and graphite peaks intensity ratio ( $\sim 0.53$ ) indicates the disordered nature of graphite. The infrared spectrum in the range 2700–3000  $\text{cm}^{-1}$  indicates the presence of  $\text{CH}_3$  asymmetric stretching, and  $\text{CH}_2$  symmetric and asymmetric stretching modes due to aliphatic hydrocarbons. This study has strong implications in understanding of the organic compounds in extra-terrestrial materials.

**Keywords:** Sadiya meteorite, Raman, infrared, organic compound.

## 1 Introduction

Meteorites are an important source of extraterrestrial matter, their chemical and physical characteristics, texture and internal structure contribute to our understanding of the birth and early history of our solar system [1]. The primitive meteorites contain abundant amounts (up to 1500 ppm) of nanodiamonds [2, 3]. The main formation process of the meteoritic nanodiamonds may be either in chemical vapour deposition or in shock origin (e.g., [4]). These nanocrystalline carbon materials (presolar grains) have to be considered amorphous or glassy carbon, and are either inorganic or consist of hydrogenated aromatic macromolecular organic carbon in the Raman spectroscopic literatures (e.g., see [5-8]). Alternatively, the carbon in primitive chondritic meteorites is present as organic material in the form of solvent –soluble and –insoluble components [9, 10]. Raman spectroscopy is a non-destructive technique that can yield information on the structural order of polyaromatic organic matter when applied at low laser power. Therefore, it is used to investigate insoluble organic matter from a range of chondritic meteorites, and a suite of interplanetary dust particles [11-13]. Raman analysis of insoluble organic matter has been already discussed by various authors in different meteorites [14-17], as well as of interplanetary dust particles [18-20] and returned samples from the Stardust Mission [21, 22]. The organic compounds in the dense interstellar clouds exhibit infrared absorption features at  $\sim 2880 \text{ cm}^{-1}$  or  $3.4 \mu\text{m}$  [23,24]. The existence of similar features (at  $3.4 \mu\text{m}$  interstellar band) in Dergaon meteorite by infrared spectroscopy has been discussed by Saikia et al. [10].

This study demonstrates the presence of organic tracers in Sadiya LL5 ordinary chondrite which fell at Natun Baliyan village of Sunpura, Sadiya ( $27^{\circ}50'09''\text{N}$ ;  $95^{\circ}51'34''\text{E}$ ), India, on 5<sup>th</sup> June 2017 (04:30 pm IST)[25] using Raman and infrared spectroscopic technique. Commonly Organic carbon including PAH are found in carbonaceous meteorite. However the organic carbon in ordinary chondrite is very rare. In our study we present our new results on occurrence of organic carbon in ordinary chondrite.

## 2 Experimental

All the sample preparation was performed in ultra-clean conditions. To prevent from the environmental artefact contaminations, such as mud, the sample was carefully checked by optical microscopy. To avoid any surface contamination and the fusion crust we fragmented the sample ( $\sim 20 \text{ mg}$ ) and took only pieces

coming from its interior. The Raman spectra were collected on bulk powdered meteorite samples using an Ar ion laser with a power of  $\sim 5$  mW, which used an excitation source having wavelength 488 nm coupled with a Jobin-Yvon Horiba LabRam-HR Micro-Raman spectrometer equipped with an Olympus microscope with 10X, 50X and 100X objectives, using the method described elsewhere [26, 27]. The powdered sample was homogenized in spectrophotometric grade KBr (1:20) in an agate mortar and was pressed 3mm pellets with a hand press. We tried to minimize the grinding time to avoid the deformation of the crystal structure, the ion exchange and the water absorption from atmosphere. The infrared spectra was acquired using Perkin-Elmer system 2000 FTIR spectrophotometer with helium-neon laser as the source reference, at a resolution of  $4\text{ cm}^{-1}$ . The spectra were taken in transmission mode in the region  $400\text{--}4000\text{ cm}^{-1}$ . The room temperature was  $29^\circ\text{C}$  during the experiment. The  $800\text{--}1100\text{ cm}^{-1}$  ( $10\mu\text{m}$ ) and  $800\text{--}400\text{ cm}^{-1}$  ( $20\mu\text{m}$ ) region of the infrared spectra facilitate an understanding of the relative nature of the  $\text{SiO}_4$  tetrahedra. Particular interest is directed towards the presence of trace of organic compound in the spectral region around  $2800\text{--}3000\text{ cm}^{-1}$  ( $3.4\mu\text{m}$ ), due to the aliphatic functional  $\text{CH}_2$  and  $\text{CH}_3$  groups.

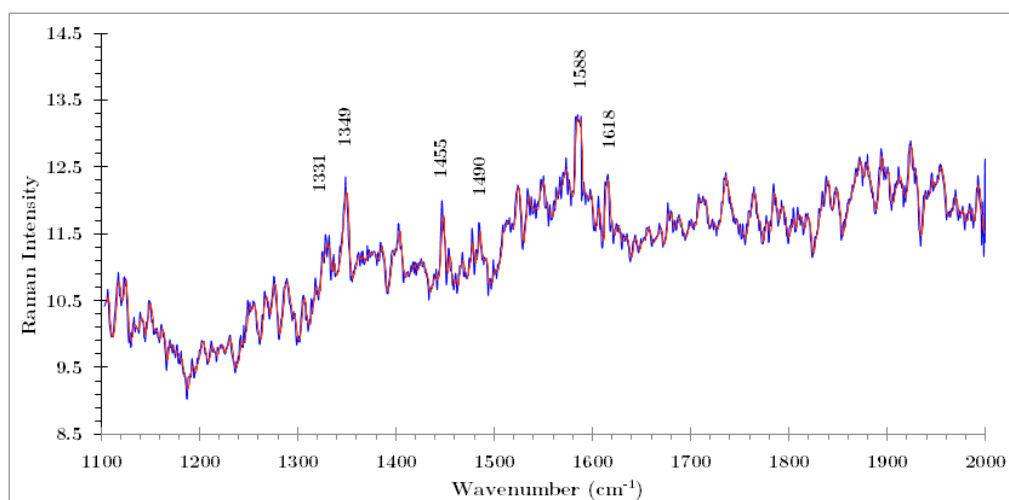
### 3 Results and Discussion

The compositional, mineralogical and Raman spectroscopic analyses of Sadiya meteorite have been discussed by Saikia et al. elsewhere (see [25]). The olivine and pyroxene composition of Sadiya meteorite are determined as:  $\text{Fa}_{28.97}$ ;  $\text{Fo}_{71.03}$ ;  $\text{Fs}_{24.47}$ ;  $\text{En}_{74.03}$  and  $\text{Wo}_{1.5}$ . The Raman spectra exhibit two characteristics peaks of olivine at  $820\text{ cm}^{-1}$  and  $850\text{ cm}^{-1}$  and three characteristic peaks of pyroxene at  $335\text{ cm}^{-1}$ ,  $679\text{ cm}^{-1}$  and  $1004\text{ cm}^{-1}$ . The olivine and pyroxene phases have already been discussed by the author in [25], therefore, the emphasis is given to the carbon phases in the Raman spectra of the meteorite.

Figure 1 displays different carbon phases observed in Raman spectra in between  $1100\text{--}2000\text{ cm}^{-1}$ . In general, the Raman spectra of nanodiamonds reveal two broad bands centred at  $1326\text{ cm}^{-1}$  and  $1590\text{ cm}^{-1}$  [26]. The first-order Raman band ( $F_{2g}$ ) of diamond occurring at  $\sim 1332\text{ cm}^{-1}$  corresponds to carbon  $sp^3$  bonding (the main C–C bond vibration in diamond) and the band ( $E_{2g}$ ) at  $\sim 1590\text{ cm}^{-1}$  is assigned to carbon  $sp^2$  bonding graphitic structures [28]. The peaks exhibiting in the Raman spectrum (Fig.1) at  $1331\text{ cm}^{-1}$ ,  $1349\text{ cm}^{-1}$  and  $1588\text{--}1618\text{ cm}^{-1}$  attributed to diamond and graphite (D and G carbon bands), respectively. The peak intensity is generally used as a signature of crystalline quality. The relative intensities of these two bands reflect the degree of ordering [29]. The peak intensities of these bands are relatively low, which indicate strong background fluorescence. Depending on the degree of disordering and orientation of the graphite grain, a second band around  $1350\text{ cm}^{-1}$  and a shoulder at  $1620\text{ cm}^{-1}$  of lower intensity is present [30], which can be observed in the Raman spectrum of Sadiya meteorite at  $1349\text{ cm}^{-1}$  and  $1618\text{ cm}^{-1}$ . The relative intensities of the  $1349\text{ cm}^{-1}$  and  $1588\text{ cm}^{-1}$  peaks reflect the degree of ordering or disordering [29]. The Raman peak position  $1618\text{ cm}^{-1}$  is the evidence of disordered graphite in the Sadiya meteorite (Fig. 1).

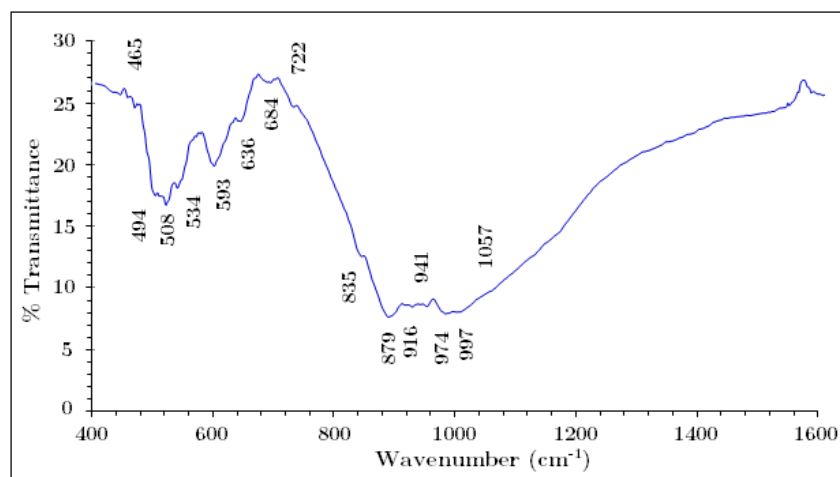
The peak appearing at  $1331\text{ cm}^{-1}$  is accepted for diamond, such Raman peaks at  $\sim 1318\text{--}1333\text{ cm}^{-1}$  for apparently shock-produced diamond inside graphite nodules from Canyon Diablo were found by Miyamoto [31]. Diamond formation by chemical vapour deposition at low pressures and high temperatures is also possible in stellar atmospheres where the nanodiamonds of chondrites [32] were generated. The artificially produced chemical vapour deposited nanocrystalline diamonds exhibit two significant bands at  $1150\text{ cm}^{-1}$  and  $1450\text{ cm}^{-1}$  [33]. Similar peak at  $1455\text{ cm}^{-1}$  is observed in Sadiya spectra. Therefore, it may be believed that the formation of nanodiamonds in Sadiya is similar to this process (chemical vapour deposition).

In general, FWHM  $10\text{--}120\text{ cm}^{-1}$  is characteristic for shock-induced diamonds [31]. We measured FWHM value for Sadiya as  $17.5\text{ cm}^{-1}$  which reflects the nature of shock metamorphism in the meteorite samples. The intensity ratio of the Raman diamond ( $I_D$ ) and graphite ( $I_G$ ) peaks in graphitic carbon was attributed by Wopenka et al. and ranging for fairly well ordered graphite ( $I_D/I_G < 0.5$ ), disordered graphite ( $0.51 < I_D/I_G < 1.1$ ), and glassy carbon ( $I_D/I_G > 1.1$ ) [22, 34]. We measure the intensity ratio of diamond and graphite peaks of Sadiya and found to be  $\sim 0.53$ . It implies to the evidence of disordered nature of graphite in Sadiya. The peaks at  $1455\text{ cm}^{-1}$  and  $1490\text{ cm}^{-1}$  are related to phonon modes with  $q_0$ , which are activated by the disorder induced by small grain sizes in nanocrystalline or amorphous diamond [35].



**Figure 1.** Raman spectrum of graphite nodule in Sadiya meteorite sample showing two peaks, at  $1349\text{ cm}^{-1}$  and peak at  $\sim 1588\text{--}1618\text{ cm}^{-1}$ . The diamond peak appears at  $1331\text{ cm}^{-1}$  (see Appendices).

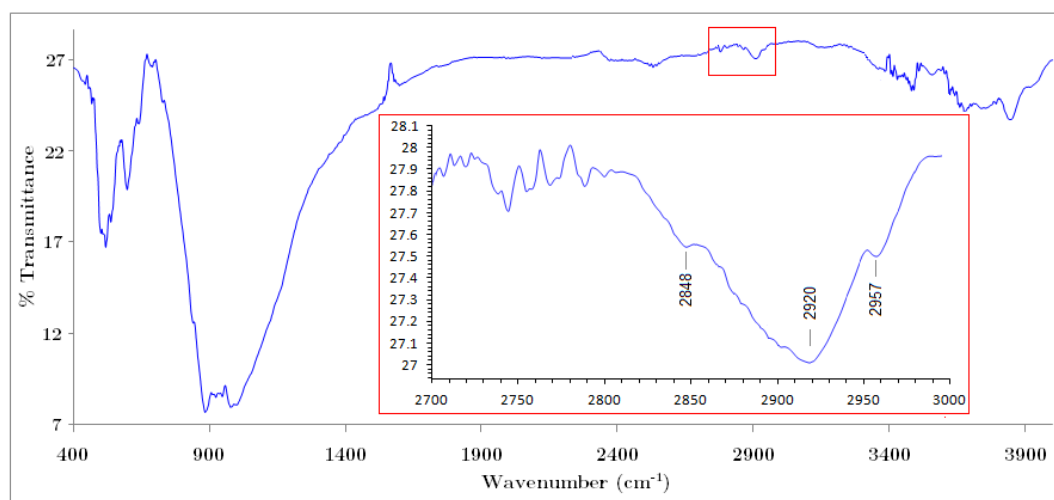
The infrared spectrum of Sadiya (Fig. 2) reveals numerous absorption bands in  $400\text{--}1400\text{ cm}^{-1}$  indicating the presence of silicates. The observed band at  $508\text{ cm}^{-1}$  can be interpreted as Si-O and Mg-O vibration modes in enstatite ( $\text{MgSiO}_3$ ) with slight shifts in the matrix [36, 37]. Generally, below  $500\text{ cm}^{-1}$  (e.g. the observed peaks between  $465\text{--}494\text{ cm}^{-1}$ ) olivines containing different divalent cations ( $\text{M-SiO}_4$ , where M is  $\text{Mg}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Mn}^{2+}$ , or  $\text{Ca}^{2+}$ ) exhibit a fairly uniform infrared spectrum in which individual bands are sensitive to cation composition [27]. The Si-O asymmetric stretching vibration (TO2-T2O5) is observed in between  $974\text{--}1057\text{ cm}^{-1}$ . The peaks arise in between  $916\text{--}974\text{ cm}^{-1}$  is assigned to the Si-O asymmetric vibration (TO3), whereas the peak arises at  $879\text{ cm}^{-1}$  due to (T2O7-TO4). The appearance of these bands is due to the Si-O mode originating from different silicate (mostly meta- and ortho-silicates) present in the meteorite. The observed bands at  $684$ ,  $722$ ,  $874$ ,  $918$ , and  $1057\text{ cm}^{-1}$  can be attributed to pyroxene.



**Figure 2.** The infrared spectra of Sadiya meteorite in the spectral region  $400\text{--}1600\text{ cm}^{-1}$ , silicate is recorded in  $400\text{--}1000\text{ cm}^{-1}$  region.

The organic aliphatic C-H stretching features in meteorites have been observed by many authors (e.g. [38-40]). During their studies, three major peaks are found in the  $2700\text{--}3000\text{ cm}^{-1}$  region, corresponding to aliphatic hydrocarbon stretching features. The pair of peaks at  $2922\text{ cm}^{-1}$  and  $2851\text{ cm}^{-1}$  corresponds to the asymmetrical and symmetrical stretching vibrations of  $\text{CH}_2$  in an aliphatic hydrocarbon. The peaks at  $2958\text{ cm}^{-1}$  and  $2865\text{ cm}^{-1}$  correspond to the asymmetrical and symmetrical stretching vibrations

of  $\text{CH}_3$  also in an aliphatic hydrocarbon [38-40]. The infrared spectrum of the Sadiya meteorite in the range  $2700\text{-}3000\text{ cm}^{-1}$  is shown in Fig. 3. The observed peaks at  $2920\text{ cm}^{-1}$  and  $2848\text{ cm}^{-1}$  correspond to the asymmetrical and symmetrical stretching vibrations of  $\text{CH}_2$ , whereas the peak at  $2957\text{ cm}^{-1}$  arises due to the asymmetrical stretching vibrations of  $\text{CH}_3$  [10]. The C-H bending vibrations of aliphatic hydrocarbons are generally observed at  $\sim 1466\text{ cm}^{-1}$  and  $\sim 1379\text{ cm}^{-1}$ , but these peaks are not observed in the present analyses. Extraterrestrial organic compounds present in carbonaceous chondrites known to exhibit large structural diversity, which is a feature contrasting with terrestrial biogenic organic compounds [41]. Potiszil et al. have carried out high pressure spectroscopic measurements on organic materials and suggested that the formation of the hydrogen bonds has implications for the origination of macromolecular organic matter (MOM) in the extraterrestrial materials. Such high-pressure-induced hydrogen-bond formation is likely a process by which aromatic MOM precursors could have cross-linked to generate the organic polymers found within extraterrestrial bodies [42].



**Figure 3.** The infrared spectra of Sadiya meteorite in the spectral region  $400\text{-}4000\text{ cm}^{-1}$ , C-H features is recorded in  $2800\text{-}3000\text{ cm}^{-1}$  region (Inset).

## 4 Conclusion

The Raman and infrared spectroscopic results provide the first evidence for the possible existence of organic compounds in Sadiya meteorite. The Raman spectra exhibit the diamond and graphite (D and G carbon bands) peaks correspondingly observed at  $1331\text{ cm}^{-1}$ ,  $1349\text{ cm}^{-1}$  and  $1588\text{-}1618\text{ cm}^{-1}$ . The peak intensity of these peaks reflects the degree of ordering. The Raman peak position  $1618\text{ cm}^{-1}$  is the evidence of disordered graphite in the Sadiya meteorite. The peak appears at  $1331\text{ cm}^{-1}$  is accepted for diamond. The presence of  $1455\text{ cm}^{-1}$  in Raman spectra indicates that the formation of nanodiamonds in Sadiya is similar to the chemical vapour deposition process. The full wave at half maximum value  $17.5\text{ cm}^{-1}$  reflects the shock metamorphism in the meteorite samples. The diamond and graphite peaks intensity ratio ( $\sim 0.53$ ) indicates the disordered nature of graphite in Sadiya. The infrared analysis is consistent to the Raman spectroscopic results. The weak absorption bands found in  $2700\text{-}3000\text{ cm}^{-1}$  of Sadiya are indicative to organic compounds. Infrared spectra indicate the presence of aliphatic hydrocarbons in the Sadiya meteorite. Similar organic features have been observed by Lawless et al. and Saikia et al. in H type ordinary chondrites [10, 43].

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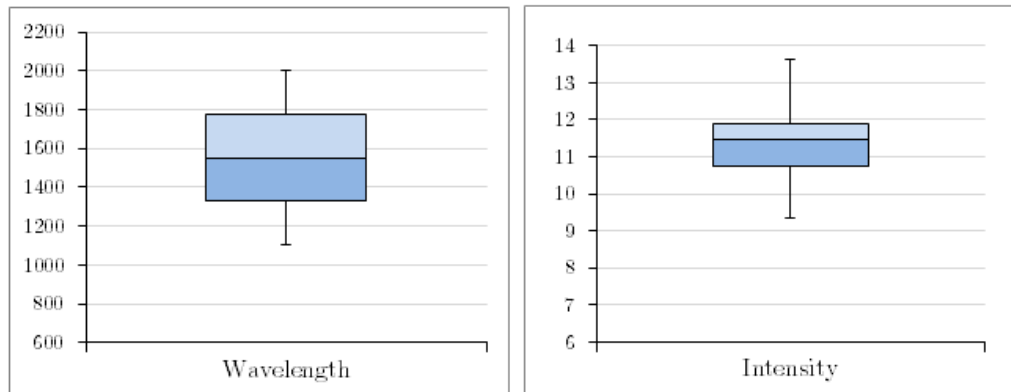
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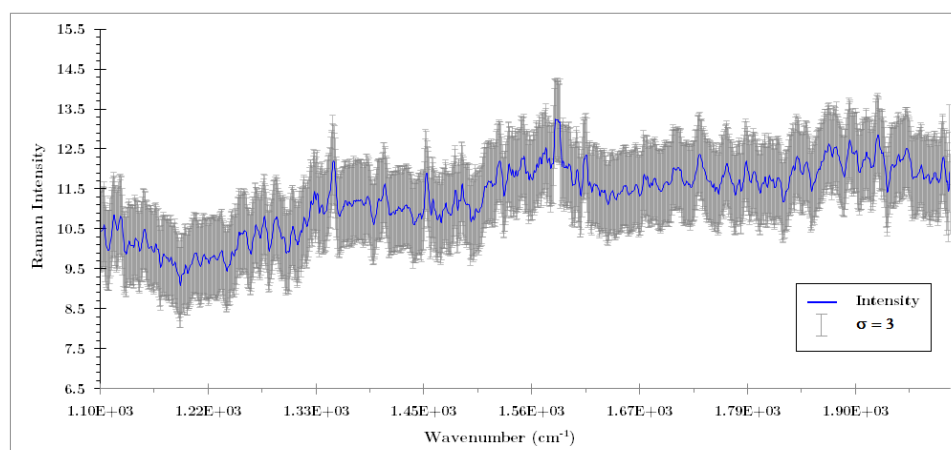
## Appendix 1:

Box-whisker plot for wavelength and intensity of the Raman spectra of Sadiya meteorite which is shown in Figure 1.



	Wavelength	Intensity
Min	1103.505127	9.02513657
Q1	1327.628815	10.7398189
Median	1552.037109	11.4648919
Q3	1775.876312	11.9008698
Max	2000.000000	13.2854107
Box 1 -Hidden	1327.628815	10.7398189
Box 2 -Lower	224.4082947	0.72507303
Box 3-Upper	223.8392029	0.43597794
Whisker Top	224.1236877	1.3845409
Whisker Bottom	224.1236877	1.7146823

## Appendix 2:



This figure represents the variation of intensity of Raman spectra (Figure 1) at standard deviation ( $\sigma=3$ ).